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Original Article

Zataria multiflora Essential oil Prevent Iron Oxide Nanoparticles-induced Liver Toxicity in Rat Model

Hamid Reza Attaran^{1*}, Faezeh Fatemi², Azadeh Rasooli³, Abolfazl Dadkhah⁴, Mohammad Reza Mohammadi Malayeri⁵ and Salome Dini⁶

¹Department of Medicine, Faculty of Medicine, Qom Branch, Islamic Azad University, Qom, Iran

²Materials and Nuclear Fuel Research School, Nuclear Science and Technology Research Institute, Tehran, Iran

³Department of Biochemistry, Faculty of Sciences, Payame Noor University, Tehran, Iran

⁴Department of Medicine, Faculty of Medicine, Qom Branch, Islamic Azad University, Qom, Iran

⁵Department of Pathobiology, Faculty of Veterinary Medicine, Garmsar Branch, Islamic Azad University, Garmsar, Iran

⁶Young Researchers and Elite Club, Karaj Branch, Islamic Azad University, Karaj, Iran

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Abstract

Over loading of iron oxide nanoparticles can causes the liver injury through overproduction of free radicals. Zataria multiflora Boiss. (Lamiaceae) has been used for many years in folk medicine due to its antioxidant and antibacterial activities. This study evaluates -for the first time- the effect of Z. multiflora essential oil (EO) against iron oxide nanoparticles hepatotoxicity in rat model. Male Wistar rats were divided randomly into 4 groups; control group (C), non-treated control group (NT), treatment groups, nanoparticles plus Z. multiflora EO at 100 and 200 mg/kg b.w. In following, the liver and blood tissue were collected for estimating oxidative stress/antioxidant parameters including lipid peroxidation (LP), glutathione (GSH), glutathione-S-transferase (GST), and liver enzyme asparate transaminase (AST) and also histopathological examinations. Iron oxide nanoparticles toxicity produced a significant increase in the levels of LP, ferric reducing ability of plasma (FRAP), and AST activity concomitant with decrease in the levels of GSH and GST activity compared to the control group. However, in the treatment groups received Z. multiflora EO, the levels of LP and AST activity reduced together with increasing in GSH content and GST activity were significantly reported. Histopathological studies also supported the biochemical assessments indicating hepatic improvements induced by nanoparticles. Z. multiflora EO protected the liver from injuries induced by iron oxide nanoparticles which can be correlated with its antioxidant activities. This can be a valuable candidate in modulating the oxidative hepatic injuries.

Keywords: Hepatotoxicity, Oxidative stress, Antioxidant parameters, Antioxidant agents

Introduction

A huge mass of biochemical functions and physiological roles of the liver makes it a crucial factor of health and a target in disease. Acute hepatic injury known as the most serious health problem worldwide is mainly caused by exposing to hepatotoxins, xenobiotics, and chemotherapeutic agents that lead to impairment of its functions [1]. Nanoparticles with very special feature (shape, size, and high proportion of surface to volume) including the elements of cobalt, iron, and nickel can easily enter into bloodstream thorough respiration and digestion systems [2]. Therefore, the liver can be seen as a target organ to identify

*Corresponding author: Department of Medicine, Faculty of Medicine, Qom Branch, Islamic Azad University, Qom, Iran Email Address: attaranhamidreza@yahoo.com

adverse effects of *in vivo* exposure to iron oxide nanoparticles.

The liver damage due to iron nanoparticles is wellinformed [3]. Iron oxide nanoparticles can be used for medical imaging, disease diagnosis, drug delivery, cancer treatment, gene therapy, and other cases. These particles accumulate in the liver cells and lead to oxidative stress with generation of reactive oxygen species (ROS) [3]. The role of iron overloading in the progression of hepatic damage in various clinical and experimental conditions has usually been studied by iron loading [4-6]. Cell culture studies confirm the toxicity of engineered nanoparticles reporting cytotoxicity, decreased cell viability, and the production of proinflammatory agents [7-9]. These cell culture studies indicate that the size and particle composition can dramatically modify toxicity [8-10].

Recently, the hepatoprotective activity of several medicinal plants with fewer side effects has been reported [11,12]. One study [13] demonstrated that silymarin and deferoxamine have a protective effect on iron-induced hepatotoxicity in rats. Silymarin can also inhibit lipid peroxidation (LP) with a significant increase in the cellular antioxidant defense machinery by ameliorating the deleterious effects of free radical reaction and the increase in glutathione (GSH) content [14]. Another study showed that [15] the hepatoprotective and antioxidant properties of aqueous fruit extract of Embilica officinalis (Euphorbiaceae).

Zataria multiflora Boiss. (Lamiaceae), commonly called Avishan-e-Shirazi, is native to central and southern parts of Iran [16]. Although, the herb is used as food flavoring, it is diversely utilized in the traditional medicine for its antiseptic, analgesic, antispasmodic, and anti-inflammatory properties [17-19]. Our previous GC and GC/MS analysis of Iranian Z. multifliora essential oil (EO) identified the main bioactive compounds which were thymol, carvacrol, p-cymene, and -terpinene. The results also have been informed the direct correlation between those major compositions and high antioxidant and antimicrobial potentials of the We also demonstrated that the essential oil. chemopreventive activity of Iranian Z. multiflora EO against DMH-induced colon carcinogenesis through modulatory DMH-metabolizing enzyme activities, i.e. cytochrome P450 (CYP450) and glutathione-S-transferase (GST), concomitant with decreasing levels of -catenin protein level [22].

Other studies also indicated a great range of *Z. multiflora* EO biological activities such as antifungal, anti-inflammatory [18,23-25], and antimicrobial, anti-nociceptive, antierythma, antibacterial [26-35] which could be attributed to the presence of high quantity of carvacrol and thymol as main phenolic compounds and p-cymene as the main non-phenolic compounds [36].

To the best our knowledge, the protective effect of *Z. multiflora* EO has not reported yet. So, in order to propose an alternative drug which would be used more safely and efficiently in the liver injuries as well as following our recent achievements, in this study, we evaluate the hepatoprotective activity of Iranian *Z. multiflora* EO against iron oxide nanoparticles induced liver toxicity in rat model.

Material and Methods

Preparation of Z. multiflora EO

Fresh Z. multiflora was collected in May 2010 from the Shiraz City (Latitude: 29°30'N Longitude: 55°00'E), Iran. Dr. Younes Asri (Botanist) authenticated the plant materials from herbarium of Iranian botanical garden (TARI) (Voucher Number: 41754).

EO Extraction

Oil extraction from the aerial parts of *Z. multiflor*a was carried out using a Clevenger-type apparatus. The extraction was carried out for 2 h, and the oil was stored in dark glass bottles in a freezer until further use [20].

Synthesis of Iron Oxide Nanoparticles

Alpha- Fe_2O_3 iron oxide nanoparticles were synthetized by co-perception method in which $FeCl_3$ 6H20 was used as a precursor and ammonia as precipitant. Morphology, size distribution and zeta potential of the synthetized nanoparticles were characterized by TEM and DLS techniques, respectively. Type of the synthetized nanoparticles was characterized by XDR technique. The size of nanoparticles was 20 nanometers. It should be mentioned that the nanoparticle did not have any charges.

Animal Treatments

Male Wistar rats 3-4 months of age, weighing 100 g were used throughout this study. The animals were obtained from the Pasteur Institute of Iran and maintained in the animal house facilities. They

were maintained on a commercial pellet food and tap water *ad libitum*. The animals were divided into 4 groups (n=5). In control group (C), rats received normal saline (nanoparticles solution) and dimethyl sulfoxide (DMSO) daily for 3 days. In non-treated control group (NT), iron oxide nanoparticles (Fe₂O₃) (200 mg/kg b.w) was i.p injected daily for 3 days. In the treatment groups, iron oxide nanoparticles (200 mg/kg b.w) plus *Z. multiflora* EO at 100 and 200 mg/kg b.w were i.p injected daily for 3 days. At day 4th, rats were anesthetized and the blood was collected by heart puncture. Rats were killed and liver were removed and processed for histological and biochemical assays as described below.

Preparation of Tissue Homogenate and Plasma

The heparinated blood samples were collected at 4^{th} day by heart puncture from all the animals and centrifuged at 3000 *g* for 10 min to obtain the plasma. Liver samples were immediately transferred to ice-cold containers and homogenized (20%, w/v) in the appropriate buffer using a homogenizer (E.L.M 2500). The homogenates were used to measure the biochemical parameters.

Biochemical Assays

CYP450 protein assay

CYP450 protein level was performed by ELISA on liver preparations according to the procedure described in the kit from Bioassay Technology Laboratory, China.

GST Activity

Liver cytosolic GST activities were measured spectrophotometrically using CDNB as substrate as described by Habig *et al.* (1974) [37]. The specific activity was calculated based on the nmol/min/mg protein in samples which was measured by Bradford assay [38].

GSH Estimation

GSH was estimated in tissue homogenates according to the procedure of Seldak and Lindsay (1968) [39].

LP Estimation

A weighed portion of liver was homogenized in phosphate buffer (100mM, pH 7.0) and used to measure the level of thiobarbituric acid reactive substances (TBARS) as indices for lipid peroxidation. The concentration of TBARS was measured spectrophotometrically using TBA reagent based on the procedure described by Buege and Aust.

Ferric reducing ability of plasma (FRAP) assay

This assay was performed using 2, 4, 6-tripyridin-2-yl-1, 3, 5-triazine (TPTZ) reagent as described by Benzie and Strain (1996) [40]. FRAP level was calculated by plotting a standard curve of absorbance against μ mol/L concentration of Fe (II) standard solution.

Determination of Enzymatic Antioxidants

The activities of superoxide dismutase SOD and CAT were estimated in liver homogenate using the commercial kits (BioVision, Inc., USA), following the instructions given by the company.

Liver Function Tests

To confirm the liver function and injury, serum alanine transminase (ALT), asparate transminase (AST), and alkaline phasphatase (ALP) were determined spectrophotometrically according to the procedure described in the kit purchased from the Pars Azmoon, Co, Iran.

Histological Studies

Preparation of tissues and staining

The liver specimens of all animals were fixed in 10% buffered neutral formalin solution. Then, they processed routinely for paraffin embedding and sectioned for hematoxylin and eosin staining and Lillie's method as iron staining.

Lillie's method staining [41]

The sectioned liver tissues were deparaffinized and hydrated to distilled water. For ferric iron accumulation staining, the sections were placed in potassium ferrocyanide solution for 1 h. Then, they were washed well in 1% aqueous glacial acetic acid and rinsed in distilled water. Finally, the sections were dehydrated in 95% alcohol, absolute alcohol cleared in xylene and mounted. Iron and accumulations were visible as dark Prussian blue. Histopathological scoring In histopathological assessment, nuclear chromatin clumping, iron accumulation and Kupffer cells hyperplasia were the prominent changes. The histopathological changes were graded numerically to be comparable statistically. The histological scoring was performed as described by Eidi et al. (2011) [74], as followings: score 0=no visible cell changes, score 1=focal hepatocyte change on less than 25% of the tissue, score 2=focal hepatocyte change on 25-50% of the tissue, score 3=extensive, but focal hepatocyte change, score 4=global hepatocyte change.

Statistical Analysis

Data are presented as means \pm Standard Error of Mean (SEM) of five samples obtained from five animals in each group. The results were subjected to one-way ANOVA followed by Tukey's HSD using SPSS (version 19.0) software. Significant levels were defined as P<0.05. (*) denote significantly different from the respective negative control group (P<0.05). (**) denote significantly different from the respective control group (P<0.05).

Results

The effects of *Z. multiflora* EO on the oxidative stress/antioxidant parameters and metabolizing enzymes in rats treated by iron oxide nanoparticles

The results indicated that LP level (Table 1) increased as evident by significant rise in the level of MDA (Table 1) at 72 h after iron oxide nanoparticles administration (P<0.05). The iron oxide nanoparticles administration also caused the liver GSH depletion at 72 h after the treatment (P<0.05) (Table 1). The LP and GSH levels returned significantly (P<0.05) to the normal levels after using Z. multiflora EO compared to the positive group at 72 h (Table 1). Since iron oxide nanoparticles, as a hepatotoxic agent, induced oxidative damage, the levels of liver antioxidant enzymes SOD, CAT and also plasma FRAP were measured (Table 1). There were no significant changes in CAT and SOD activities (Table 1) after 72 h of iron oxide nanoparticles administration. While, the abnormal high level (P<0.05) of FRAP observed in the iron oxide nanoparticle induced liver toxicity at 72 h after nanoparticles treatment (Table 1). The serums of FRAP, SOD, and CAT levels did not show any remarkable changes after treatment of *Z. multiflora* EO (Table 1). As shown in Table 1, CYP450 protein level did not significantly alter in the liver of all groups. The level of the detoxifying enzyme GST (Table 1) diminished in rats treated with toxicity dose of iron oxide nanoparticles at 72 h (P<0.05). While, the administration of *Z. multiflora* EO at 100 and 200 mg/kg b.w surprisingly elevated the reduction of serum GST activity induced by iron oxide nanoparticles (Table 1).

The effects of *Z. multiflora* EO on the liver enzymes in rats treated by iron oxide nanoparticles

Serum biochemical parameters including AST, ALT, and ALP in all experimental groups were depicted in Table 2. As shown in Table 2, the activity of the serum AST increased markedly (P<0.05) in iron oxide nanoparticles-treated rats at 72 h as compared to the NT group. However, coadministration of rats with Z. multiflora EO surprisingly (P<0.05) restored iron oxide nanoparticles induced the elevation of the serum AST level (72 h) towards normal value (Table 2). Also, the ALT and ALP levels did not affect by treating with iron oxide nanoparticles in all groups (Table 2).

Histopathological Assessments

Histopathologic analysis did not show any obvious lesion in the liver of control group. Iron staining did not show any iron accumulation as well (Fig. 1A and Table 3).

Table 1 The effects of Zataria multiflora Boiss. EO on the oxidative stress/antioxidant parameters and metabolizing enzymes in the rats treated by iron oxide nanoparticle at 72 h

| Groups | LP (n mol/mg protein) | GSH (n mol/mg protein) | FRAP (µmol/L) | CAT (mU/mg protein) | SOD (mU/mg protein) | CYP450 (pg/mg protein) | GST (n mol/min/mg protein) |
|--------|-----------------------------|----------------------------|------------------|---------------------------|---------------------------|---------------------------|----------------------------------|
| С | 1.39±0.17 | 460.4±13.03 | 631.60±40.37 | 7.26±0.35 | 12.06±0.75 | 1722±91.29 | 784.4±47.33 |
| NT | 2.53±0.11* | 342.4±19.2 [*] | 436.40±17.45* | 6.63±0.27 | 11.79±0.61 | 1668±77.09 | $509.8 \pm 28.92^*$ |
| EO 100 | 1.45±0.11** | 435.25±28.28 ^{**} | 445.20±15.10 | 6.88±0.37 | 12.02±0.78 | 1670±60.13 | 771±35.29** |
| EO 200 | 1.41±0.15** | 459.75±15.83** | 412.25±22.27 | 7.19±0.38 | 12.11±0.77 | 1662.5±102.01 | 819±55.32** |

C: control group; NT: non-treated control group; EO: *Zataria multiflora* Boiss.essential oil (100 & 200 mg.kg b.w) treated groups. $^*P<0.05$ is considered significantly between control group (C) and non-treated control group (NT). $^{**}P<0.05$ is considered significantly between non-treated control group (NT) and *Z. multiflora* EO treated groups. Data are presented as mean \pm SD.

| Groups | AST (U/L) | ALT (U/L) | ALP (U/L) | |
|--------|--------------------------|------------|-------------|--|
| С | 112±7.26 | 37.80±2.63 | 360±20.24 | |
| NT | $125.60{\pm}8.38^{*}$ | 39.40±3.55 | 384±22.04 | |
| EO 100 | 121.20±8.86** | 36.25±2.05 | 363.7±23.57 | |
| EO 200 | 113.5±7.68 ^{**} | 34±1.95 | 366.2±19.08 | |

Table 2 The effects of Zataria multiflora Boiss. EO on the liver enzymes in the rats treated by iron oxide nanoparticle at 72 h

C: control group; NT: non-treated control group; EO: *Zataria multiflora* Boiss. essential oil (100 & 200 mg.kg b.w) treated groups. $^*P<0.05$ is considered significantly between control group (C) and non-treated control group (NT). $^{**}P<0.05$ is considered significantly between non-treated control group (NT) and *Z. multiflora* EO treated groups. Data are presented as mean \pm SD.

Table 3 Analysis of liver histopathological changes in different study groups

| Groups | С | NT | EO (100 mg/kg b.w) | EO (200 mg/kg b.w) |
|---------------------------|------|-----------|-----------------------|-----------------------|
| Iron accumulation | 0±0 | $4\pm0^*$ | 2.6±0.5 | 1.3±0.5 ^{**} |
| Chromatin clumping | 0±0 | 3±0* | 0.3±0.5 ^{**} | $0.3\pm 0.5^{**}$ |
| Kupffer cells Hyperplasia | 0± 0 | 3.3±0.5* | 2.6±0.5 | 1.3±0.5 ^{**} |

C: control group; NT: non-treated control group; EO: *Zataria multiflora* Boiss. essential oil (100 & 200 mg/kg b.w) treated groups. $^*P<0.05$ is considered significantly between control group (C) and non-treated control group (NT). $^{**}P<0.05$ is considered significantly between non-treated control group (NT) and *Z. multiflora* EO treated groups. Data are presented as mean \pm SD.

On the other hand, iron oxide nanoparticles significantly accumulated in the periportal hepatocytes and phagocytic cells (Kupffer cells) in the liver of NT group due to nanoparticles injection in comparison with control group (P<0.05).

Increased number of phagocytic cells, clumping of nuclear chromatin and apoptosis of hepatocytes were observed in the periportal region of nanoparticles treated group (NT Group) (Fig. 1B and Table 3). *Z. multiflora* EO administration reduced iron accumulation and the number of Kupffer cells in the liver. However, 200 mg/kg b.w EO administration was more efficient (P<0.05) in comparison with NT group (Fig. 1D and Table 3).

Discussion

Recently, there is a great demand to find a suitable hepatoprotective agent among the natural products especially medicinal plants against toxic chemicals and drugs induced liver injuries. So, the present study was designed to estimate the potential of *Z. multiflora* EO in restoration the equilibrium of the

oxidative stress/antioxidant system in the liver of rats treated with iron oxide nanoparticles.

The nanoparticles rapidly enter circulation and may migrate to the liver, kidney, spleen, lungs, and brain [42-44]. Although the mechanism which nanoparticles cause toxicity in the liver is still uncertain, one of the most probably mechanisms of toxicity was previously pointed the generation of free radical products especially reactive ROS [44]. ROS like hydroxyl radical (HO[•]), hydrogen peroxide (H_2O_2) , and superoxide anion (O^{2-}) is responsible for stimulating a variety of cellular and physiological actions such as inflammation, apoptosis, and deoxyribonucleic acid (DNA) damage [45-46]. Indeed, ROS has been known to produce cellular and tissue injury through covalent bindings such as DNA strand breaking, increase of LP, and augment fibrosis [3,47,48]. Several studies also suggested that iron oxide nanoparticles were found to significantly reduce the levels of cellular antioxidants leaded to the oxidative damages of intracellular proteins and DNA and increased LP levels [49,50].



Fig. 1 Effect of *Zataria multiflora* Boiss. EO on histopathological changes 72 h after iron nanoparticles administration. A: control group, A1; liver in control group were completely normal, H&E 400^{*}. A2; there is not any iron **a**ccumulation in the liver, Lillie's method staining 100^{*}. B: non-treated control group, B1; periportal changes consist of clumped chromatin, increased phagocytic cells as dark small nuclei (small arrows) and apoptosis (large arrows), H&E 400^{*}. B2; blue stain accumulation of iron particles in hepatocytes and Kupffer cells of periportal region (arrows), Lillie's method staining 400^{*}. C: EO treatment group (100 mg/kg b.w), C1; the hepatocytes show the normal condition, H&E 400^{*}. C2; iron accumulation has decreased in liver in compare with iron nanoparticles treated group. Lillie's method staining 400^{*}. D: EO treatment group (200 mg/kg b.w), D1; The liver is in the normal condition, H&E 400^{*}. D2; There is not any obvious iron accumulation, Lillie's method staining 400^{*}.

In addition, other effects of iron oxide nanoparticles include the reduction of GSH liver cells, increase shock, and oxidative reactions in the liver cells [5]. These changes were confirmed in our study by decreasing in the GSH content together with the elevation of LP level after nanoparticles administration as shown in Table 1. Also, the FRAP level decreasing in the NT group indicating it as an important factor in the balance between oxidative stress and antioxidant status (Table 1). Moreover, the level of AST significantly increased (P<0.05) by 200 mg/kg b.w iron oxide nanoparticles consumption as compared to the normal group (Table 2). Many studies have been done on the hepatotoxicity characteristics of nanoparticles showed that these substances has toxic effects on the liver by increasing ALT, AST and ALP activities [52]. Another study [3] indicated that the significant increase in (P < 0.05)AST, ALT, and ALP levels in the groups exposed to the over doses of nanoparticles (150 µg/kg) as compared to the normal group. Also, Jensen et al. (2003) investigated the relationship between the extent of hepatocellular injury as reflected by serum levels of ALT, AST, and several iron status indices in 39 anti-hepatitis C virus-negative patients with transfusional iron overload owing to acquired anemia. They observed that serum levels of ALT and AST were directly involved in iron toxicity [53].

Our data were supported by our histopathological examinations (Fig. 1B and Table 3) indicating that iron oxide nanoparticles accumulated in the periportal hepatocytes and phagocytic cells in the liver of NT group due to nanoparticles injection. To increase in the number of phagocytic cells, clumping of nuclear chromatin and apoptosis of hepatocytes were observed in the periportal region of nanoparticles treated group (Fig. 1B and Table Hepatic accumulation of 3). iron oxide nanoparticles affects the liver function. Also, stem cells and fibroblast cells were affected with decreasing of mitochondrial activity which leads to the morphological changes. The results of the present study are similar to the previous reports [54-56] revealed that nanoparticles such as iron oxide nanoparticles can enter into the cells and accumulated in the mitochondria, vesicles. phagosomes, and lysosomes [57-59].

Currently, there have been numerous reports in the hepatoprotective effect of the several medicinal plants [11,60]. As the variable side effects of modern medicine, the demand for herbal drugs as safe and effective substances for treating injuries is getting more attentions [61]. The phytochemicals derived from plants possess multiple activities. It has been found that natural formulated compound is more active than the isolated form [62]. We previously reported that several medicinal plants with their antioxidant properties could protect the liver against deleterious effects of hepatoxic agents in rat models [21,63-67]. Moreover, we observed that Z. multiflora EO containing high percent of carvacrol had thymol and noticeable chemopreventive, antioxidant, and antimicrobial properties in *in vivo* and *in vitro* systems [20, 64]. In followings, in this study, we demonstrated that Z. multiflora EO inhibit nanoparticles induced liver injuries as demonstrated by restoring the levels of LP (Table 1), GST (Table 1), and serum AST (Table 2). Also, the administration of Z. multiflora EO mediated (P<0.05) the reduction of GSH content induced by nanoparticles (Table 1). The histopathological findings (Figs. 1C, 1D and Table 3) confirmed that Z. multiflora EO improved the liver injury through modulating oxidative stress/antioxidant parameters. In other words, the antioxidant activity of Z. multiflora EO may be responsible for protecting the liver from toxicity. Z. multiflora EO with the major antioxidant components as thymol, carvacrol, p-cymene, and terpinene may enhance the innate mechanisms of the antioxidant system or provide its antioxidant capacity. According to the recent reports, the biological activities of Z. multiflora are mostly due to its phenolic compounds; particularly thymol and carvacrol [68,69]. One study [68] also pointed that the existence of thymol in EO might contribute to its pharmaceutical effects like anti-inflammatory, antioxidant, and anticholinesterase properties. It is well documented that thymol and carvacrol act on the microbial cell membrane and cause substantial morphological damage, resulting in a change in permeability and the release of cellular contents [70,71]. Previous studies have also shown that some EO owing scavenging free radicals and antioxidant properties possess considerable hepatoprotective activity [72,73].

Conclusion

The hepatoprotective activity of *Z. multiflora* EO against iron oxide nanoparticles induced liver injuries in *in vivo* experiment was confirmed by

modulating the various oxidative stress/antioxidant parameters *e.g.* LP, GSH, and GST as well as liver enzymes like AST. It was also proved by histopathological findings indicating *Z. multiflora* EO modulatory effects in improving oxidative liver injuries.

Declaration of Interest

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