Original Article



Identification of Essential Oil Components, Total Phenolic Content and Antioxidant Properties of *Mentha aquatica* in Southern Khorasan Province, Iran

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| Article History | ABSTRACT |
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| Received: 17 April 2021 Accepted: 08 April 2022 © 2012 Iranian Society of Medicinal Plants. All rights reserved. | Many medicinal plants have natural antioxidant properties. In this study, the chemical composition of <i>Mentha aquatica</i> L. essential oil was determined and antioxidant properties of essential oils and methanolic extracts, ethanolic extracts, acetonic extracts and aqueous extracts were studied. Also, Phenolics, flavonoids and tannins contents of all the mentioned extracts were determined. The essential oil of the dried flowering aerial part of wild water mint is extracted by hydrodistillation. The essential oil was analyzed by capillary GC and GC-MS. The Folin–Ciocalteu method was adopted to |
| Keywords Antioxidant properties <i>Mentha aquatica</i> Essential oil Methanolic extract Total phenolic content | analyzed by capitary GC and GC-MS. The Folm–Clocated method was adopted to determine the total phenols content while flavonoids were estimated according to the aluminum chloride colorimetric method. To evaluate tannins content, vanillin and HC were added to the extracts. The antioxidant potential was measured by 1, 1-diphenyl-2 picrylhydrazyl radical scavenging and the inhibition of β -carotene bleaching assays The main components of essential oil were pipritenone oxide (37.80%), 1, 8-cineole (24.13%), α -gurjunene (11.96%) and pulegone (4.64%). The antioxidant properties o the essential oil were more considerable than all the other extracts. The antioxidan properties and total phenolics, flavonoids and tannins contents of methanolic extracts |
| *Corresponding author Nakhaei90@yahoo.com | ethanolic extracts and acetonic extracts were larger than the aqueous extract. These results show that essential oils, methanolic extracts, ethanolic extracts and acetonic extracts of <i>M. aquatica</i> from Southern Khorasan of Iran have a great potential or polyphenols which can be used as a natural food preservative and antioxidant source. |

INTRODUCTION

Mentha aquatic L. belongs to the Lamiaceae family used as fresh or dried in herbal drinks and condiments for their special fragrance [1]. The essential oil extracted from plants of this family is used in pharmaceutical, food and cosmetics industries. Mentha genus has been considered for pharmaceutical properties such as the easing of maldigestion and relieving muscular cramps [2]. Water mint grows wild in damp lands in the periphery of rivers and springs and is odiferous due to its possession of essential oils. The essential oil is produced in the capillary glands on the surface of the plants' aerial parts [3].

Mohsenpour *et al.* [4] identified the ingredients of wild *M. aquatica* essence in Mazandaran Province and reported menthoforan, trans-caryophyllene, eucalyptol and germacron as the main ingredients of its essence. Bozin *et al* [5], Getahun *et al.* [6] and Sutour *et al.* [7] also confirmed the existence of

menthofuran, 1, 8-cineole and trans-caryophyllene in the essence of *M. aquatica*. Andro *et al.* [8] investigated the essence building blocks of *M. aquatica* and identified menthofuran, limonene, iedol and caryophyllene. Chaker *et al.* [9] recognized such building blocks as α -pinene, sabinene, myrcene, limonene, α -terpineol, terpinen-4-ol and pulegone in the essences of *M. aquatica* plants from Algeria. Zamfirache *et al.* [10] found limonene, α -pinene, sabinene, borneol and myrcene in the essence extracted from *M. aquatica*.

Antioxidants are compounds that can effectively prevent the oxidation of macromolecules such as lipids, proteins and nucleic acids, hence preventing the damage and programmed death of the cells as well as cardiovascular diseases and cancers [11]. The use of chemical antioxidants most frequently used in the food industry has been limited for carcinogenic and negative effects on human health [12]. Natural antioxidants that are abundant in fruits

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and vegetables have been regarded as efficacious in disease prevention. A number of these compounds are made with antioxidant properties as secondary metabolites by the wild plants amongst which such compounds as phenols, flavonoids, terpenoids and steroids can be pointed out [13]. Dhifi et al. [14] and Getahun *et al.* [6] reported that the essence of M. aquatica grown in Tunis and Ethiopia feature high antioxidant activities. Water mint grows wild in many parts of Iran including the southern Khorasan Province. Climatic conditions of southern Khorasan Province are different from other habitats of the water mint. This area is dry and the days are hot and the nights are cool in the summer. Water mint is a main medicinal plant in this region and it is also used to repel pests and harmful rodents. To date, the chemical composition of water mint essential oil has not been identified in this area. The present study aims at the identification of chemical compounds and investigation of antioxidant activities of M. aquatica essential oil and aqueous extract and alcoholic extracts, and of the phytochemical properties of the all extracts in southern Khorasan Province. Iran.

MATERIALS AND METHODS

Plant Materials

The aerial flower-bearing parts of wild *M. aquatica* L. were collected from the periphery of an aqueduct in Ghaisar Village of Khosf city in southern Khorasan Province during the summer, 2016. The plant materials were dried in the shade at room temperature for ten days. The voucher specimen of this plant is SPNH-2045 (Sari Payame Noor University Herbarium).

Essential oil was extracted and the ingredients were identified in the laboratory of Research Institute of Forests and Rangelands. In addition, other phytochemical properties were delineated in the laboratory of Islamic Azad University of Birjand.

Essential Oil Extraction

Essential oil extraction was carried out based on hydrodistillation method in the Clevenger Device. First, 100 g of the plant was used for essential oil extraction and the duration of essential oil extraction was 3 hours. The extracted essential oil was kept in a cold and dark environment in a refrigerator before being subjected to analysis [4].

Essential Oil Analysis

To isolate and identify essential oil ingredients, a gas chromatography device and gas chromatograph attached to a mass spectrometer (GC/MS) were

employed. Spectra detection was carried out based on their Kovats indices and by the injection of normal hydrocarbons (C7-C25) under identical conditions and essential oil injection using a computer program in basic language. Moreover, comparisons were made with various study sources [15-17] using a mass spectrum of standard compounds and information existent in GC/MS Device's library. Thermo-UFM gas chromatograph equipped with an FID detector was used and the data processor in Chrom-Card 2006 Software was employed for a Ph-5 column (10 m in length, 0.1mm in internal annulus and 0.4 micron in the thickness of the resident phase layer) that had been thermally programmed from 60 °C to 285 °C with a 3 °C increase per minute. The carrier gas was helium and the injection section temperature and detector temperature were set at 280 °C. The gas chromatography device was connected to the Varian 3400 Model Mass spectrometer. Saturn II was employed for analysis and the same column was utilized in the GC device. The gas pressure was 35 Psi in the column head and ionization energy was equivalent to 70 eV. The column's thermal programming was the same as that used for the GC device [4].

Preparation of Solvent Extracts

A portion (20 g) of the powdered plant material was Soxhlet-extracted with water, ethanol, acetonic and methanol for 8 h, at a temperature not exceeding the boiling point of the solvent. The extracts were concentrated using a rotary evaporator at 50 °C to obtain crude extracts. All the extracts were kept in the dark at 4 °C prior to use [18].

Determination of total phenolic content

Total phenol content was measured based on Folinciocaltue colorimetric method using gallic acid as a standard [19]. Then, about 1.8 ml of the distilled water and 0.2 ml of dilute folin-ciocaltue (1:15 v/v) reagent were added to 2 ml of plant extract or standard solution (Concentrations of 0-100 mg/ml gallic acid in the distilled water). After 5 minutes, 3 ml of 7% sodium carbonate solution was added to the reaction mixture, and the reaction mixture increased with the distilled water to 5 ml. After 90 minutes of storage at laboratory temperature, their absorption at 750 nm was examined by a spectrophotometer (Jenway, 6305) and the amount of total phenol was determined from the standard curve in terms of mg/GAE g DW.

Determination of Total Flavonoid Content

Total flavonoid content was measured by aluminum chloride colorimetric method using quercetin as 45

standard [20]. First, standard solutions with concentrations of 0-100 mg/L quercetin in absolute methanol were prepared. Then, 2 ml of the plant extract or standard solution of aluminum chloride and 1 ml of aqueous acetic acid were added and mixed well. Finally, the reaction mixture was increased to 90 ml with 90% ethanol and the tubes were stored at room temperature for 30 minutes. Their light absorption was read at 414 nm by the spectrophotometer (Jenway, 6305) and total flavonoids were obtained using a standard curvature in terms of mg/g DW.

Determination of Total Tannins Content

Total tannins were determined according to Rebaya *et al.* [21] with slight modifications. A volume of 12.5 mL of the extract was added to 750 mL of vanillin and 375 mL of HCl. The mixture was then shaken and incubated at room temperature for 15 min. The absorbance was measured at 500 nm by the spectrophotometer (Jenway, 6305) and the tannin content was expressed as milligrams of catechin equivalent per gram of dry weight (mg CE/g DW).

Antioxidant Properties Investigation DPPH Radical Scavenging Assay

A total of 50 microliters of various concentrations of the essential oil and extracts in methanol was admixed with 5 ml of 0.004 %2, and 2-diphenyl-1picrylhydrazyl (DPPH) in methanol. The specimens were placed in ambient temperature in a dark room for 30 minutes. Then, their light absorption was read in room temperature at 517 nm. Next, the following formula was utilized to calculate the percentage of the inhibition of DPPH free radicals [14]:

Inhibition percentage = (light absorption of the evidence specimen without essential oil and extractamount of the light absorption of various concentrations of essential oil and extract / light absorption of the evidence specimen without essential oil and extract) \times 100.

Then, the concentration of water mint essential oil with 50% radical inhibition (IC_{50}) was calculated by the linear regression. Obviously, smaller values mean greater antioxidant power or free radical scavenging. In this test, synthetized 2, 6-ditertbutyl-4-methylphenol (butylated hydroxyl toluene, BHT) antioxidant was used as positive control. All the experiments were repeated three times.

Carotene/Linoleic Acid Assay

A solution of β -carotene was prepared by dissolving 2 mg of β -carotene in 20 mL of chloroform. This

solution (2 mL) was mixed with 20 mg linoleic acid and 200 mg Tween 40. After the chloroform was removed at 40 °C under vacuum, 50 mL of oxygenated ultra-pure water was added, and then the emulsion was vigorously shaken. Aliquots (750 mL) of this emulsion were transferred into test tubes containing different concentrations of extracts (50 mL). After the addition of the emulsion to each tube, zero-time absorbance of the control, containing methanol instead of extract, was measured at 470 nm by the spectrophotometer (Jenway, 6305). The test samples were then incubated in a water bath at 50 °C for 120 min, when the absorbance was measured again. The β carotene bleaching inhibition was calculated using the following equation: Inhibition (%) = [(At - C / $(C0 - Ct)] \times 100$

Where, At and Ct are the absorbance values measured for the test sample and control, respectively, after incubation for 120 min, and C0 is the absorbance value for the control measured at zero time during the incubation. The results are expressed as IC₅₀ values, the concentration required to cause a 50% β -carotene bleaching inhibition. BHT was used as a positive control [22].

RESULTS

Essential oil Analysis

Table (1) provides a list of the constituents of *M*. *aquatica* essential oil and their values in percentage. The natural ingredients accounted for 96.97% (13 compounds) of the *M*. *aquatica*'s essential oil volume. Pipritenone oxide (37.80%), 1, 8-cineole (24.13%), α -gurjunene (11.96%) and pulegone (4.64%), respectively, were the basic compounds in *M*. *aquatica*'s essential oil. Such ingredients as limonene (3.76%), myrcene (3.48%), sabinene (2.93%), perilla aldehyde (2.52%), piperitone (2.07%) and α -pinene (1.96%) were found in lesser amounts.

Monoterpenes oxygenated 72.88%, monoterpenes hydrocarbons 12.13%, and sesquiterpenes hydrocarbons 11.96% were also accounted for in water mint essential oil.

Antioxidant Properties

In DPPH assay IC₅₀, the values of essential oil and BHT were 3.24 and 4.78 µg/mL respectively. The essential oil had significantly stronger antioxidant properties than methanolic extracts (IC₅₀= 25.4 µg/mL), ethanolic extracts (IC₅₀= 26.13 µg/mL), acetonic extracts (IC₅₀= 25.45 µg/mL) and aqueous extracts (IC₅₀= 51.11 µg/mL) and there was no significant difference with BHT. IC₅₀ value of all

extracts was significantly higher than BHT, so the antioxidant properties were less than the positive control. Aqueous extract had the highest IC_{50} value and therefore the lowest antioxidant properties.

Antioxidant activity determined by ßcarotene/linoleic acid emulsified system capacity was estimated by inhibiting the formation of free radicals in the β -carotene from oxidized linoleic acid. It was estimated by spectrophotometrically the paling of the orange color characteristic of βcarotene, which is related to the inhibitory capacity of free radical formation. In the Carotene/linoleic acid assay method, the antioxidant properties of the essential oil (IC₅₀ = 448 μ g/mL) were significantly higher than methanolic extracts (IC₅₀ = 665 μ g/mL), ethanolic extracts (IC₅₀ = 643 μ g/mL), acetonic extracts (IC₅₀ = 654 μ g/mL) and aqueous extracts (IC₅₀ = 971 μ g/mL). Aqueous extract had the lowest antioxidant properties and BHT had the highest antioxidant properties.

Total Phenolic, Flavonoid and Tannin Contents

All extracts had noticeable phenolic contents (Table 3). The uppermost amounts were observed for methanolic extracts (45.30 mg GAE/g DW, 34.87 mg RE/g DW and 8.77 mg CE/g DW) whereas minimum amounts were noticed for aqueous extracts (20.12 mg GAE/g DW, 15.43 mg RE/g DW and 2.32 mg CE/g DW) for the total phenolic, flavonoid and tannin contents, respectively.

Total phenol, flavonoids and tannins contents in the aqueous extract was significantly lower than other extracts. But methanolic extracts, ethanolic extracts and Acetonic extracts were not significantly different.

Table 1 Chemical ingredients constituting *M. aquatica* L. essential oil

| Row | Ingredient | CC | KI | KI ^{lit} | Percentage |
|-----|--------------------|----|------|-------------------|------------|
| 1 | α-Pinene | C1 | 937 | 932 | 1.96 |
| 2 | Sabinene | C1 | 973 | 969 | 2.93 |
| 3 | Myrcene | C1 | 988 | 988 | 3.48 |
| 4 | Limonene | C1 | 1029 | 1224 | 3.76 |
| 5 | 1,8-Cineole | C3 | 1035 | 1026 | 24.13 |
| 6 | Borneol | C3 | 1178 | 1165 | 0.36 |
| 7 | Terpinen-4-ol | C3 | 1165 | 1174 | 0.57 |
| 8 | α-Terpineol | C3 | 1172 | 1186 | 0.79 |
| 9 | Pulegone | C3 | 1235 | 1233 | 4.64 |
| 10 | Piperitone | C3 | 1251 | 1249 | 2.07 |
| 11 | Perilla aldehyde | C3 | 1270 | 1269 | 2.52 |
| 12 | Piperitenone oxide | C3 | 1357 | 1366 | 37.80 |
| 13 | α-Gurjunene | C2 | 1404 | 1409 | 11.96 |
| - | Other compound | C4 | - | - | 3.03 |

CC—chemical class; C1—monoterpenes hydrocarbons; C2—sesquiterpenes hydrocarbons; C3—monoterpenes oxygenated; C4—unidentified compounds; KI^{lit}— Kovats indices from literature.

Table 2 Antioxidant activities of essential oil and different extract of Water mint (*M. aquatica*) and Synthetic antioxidant BHT

| Antioxidant activities | Essential oil | Aqueous extract | Methanolic extract | Ethanolic extract | Acetonic extract | BHT |
|---|------------------|--------------------|-----------------------|----------------------|---------------------|-------------|
| DPPH [IC50 (µg/mL)] | 3.24±0.17 a | 51.11±1.05 c | 25.40±1.35 b | 26.13±2.11 b | 25.45±0.95 b | 4.78±0.12 a |
| β -Carotene bleaching [IC50 ($\mu g/mL$)] | 448± 0.95 b | 971±0.40 d | 665±0.70 c | 643±0.50 c | 654±0.8 c | 6.5±0.25 a |

Values are given as mean (n = 3). Values in each line followed by different letters are significantly different (P < 0.05).

Table 3 Total phenolics, flavonoids and tannins content in different extract of Water mint (M. aquatica)

| Content | Aqueous extract | Methanolic extract | Ethanolic extract | Acetonic extract |
|---------------------------------------|-----------------|--------------------|----------------------|------------------|
| Total phenolic contents (mg GAE/g DW) | 20.12±0.58 a | 45.30±0.98 b | 44.95±0.90 b | 45.15±1.05 b |
| Flavonoid contents (mg RE/g DW) | 15.43±0.30 a | 34.87±1.00 b | 34.56±0.24 b | 33.91±0.19 b |
| Tannin contents (mg CE/g DW) | 2.32±0.45 a | 8.77±0.14 b | 8.45±0.22 b | 8.24±0.12 b |

In each line, values followed by different letters are significantly different (P < 0.05).

DISCUSSION

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In the present study, pipritenone oxide and 1, 8cineole were identified as the basic ingredients of M. aquatica essential oil grown in southern Khorasan Province which is consistent with the results obtained in the studies by Esmaeili et al. [23]. They also detected pipritenone oxide and 1, 8-cineole as the fundamental compounds in the essential oil of M. aquatica collected from the Amol region. In the present study, 1, 8-cineole was identified as major constituents of water mint essential oil. Consistent with the results of the present study, Bozin et al. [5] and Senatore et al. [24] reported 1, 8-cineole as a major constituent of M. aquatica essential oil. Jerkovic and Matelic [25] and Fallahian et al. [26] also similarly reported that 1, 8-cineole is a main oxygenated monoterpene in M. aquatica essential oil.

In the present study, the antioxidant properties of M. aquatica essential oil were stronger than all other extracts. Getahun et al. [6] and Dhifi et al. [14] also investigated the antioxidant properties of the essential oil of M. aquatica grown in Ethiopia and Tunis and reported high antioxidant activities for M. aquatica (IC₅₀ values equal to 2.11 µg/ml and 8 µg/ml, in DPPH assay respectively) similar to the findings of the present study. M. aquatica plants used in the present study were grown in a hot and arid climate in the desert margin so their growth conditions were almost identical to those in Ethiopia, hence having high antioxidant properties akin to the plants grown in Ethiopia. Likewise, Grzeszczuk and Jadczak [27] reported antioxidant activities for various species of Mentha plants, especially water mint. Boumendjel et al. [28] also investigated the antioxidant properties of the essential oil of M. aquatica grown in Algeria and, similar to the findings of the present study's experiment, reported high antioxidant activities for *M. aquatica* (IC₅₀ values equal to 0.69 μ g/ml, in DPPH assay). A major oxygenated monoterpene which accounts for a large percentage of M. aquatica essential oil is 1, 8-cineole. In accordance with the present study's findings, Mimica-Dukić et al. [19] reported that the antioxidant properties of M. aquatica are predominantly the result of a large volume of 1, 8-cineol existing in the essential oil of this plant.

In the present study, we noticed that methanolic extracts, ethanolic extracts and acetonic extracts of the *M. aquatica* from Iran were rich in phenolics, flavonoids and tannins contents, thereby justifying the antioxidant ability of these extracts. Total phenolic, flavonoid and tannin contents and

antioxidant activity of aqueous extract were lower than other extracts.

According to the findings of this experiment, Benabdallah *et al.* [29] reported that the antioxidant activity of different types of mint extract was correlated with total phenolic, flavonoid and tannin contents in the extracts.

The high correlation between the values of phenols concentration in plant extracts and antioxidant activity is well documented [30, 31]. In this regard, Luximun-Ramma *et al.* [32] showed a linear correlation between antioxidant activity and phenolic contents of the plant extracts.

The antioxidant activity of polyphenols is mainly due to their ability to act as hydrogen donors, reducing agents and radical scavengers [33]. This activity is generally dependent on total phenol content [34]. In fact, the methanolic extracts, ethanolic extracts and acetonic extracts of *M. aquatica*, which had the highest phenolic, flavonoid and tannin contents, showed the greatest antioxidant activities in comparison with the aqueous extract.

Furthermore, it should be taken into consideration that antioxidant capacities might be attributed to the chemical structure of compounds, as well as synergistic or antagonistic effect of compounds present in the crude extract [35].

The essential oil and all extracts of *M. aquatica* evaluated in this study showed considerable antioxidant levels correlated to the strong polyphenols content. The present study contributed much to the application of *M. aquatica* as herbal tea or additive in foods and folk medicine for the therapy of infectious diseases.

CONCLUSION

Piperitenone oxide, 1,8-cineole and α -gurjunene were the main constituents of water mint essential oil grew in southern Khorasan Province of Iran, and the essential oil of water mint can be considered as a source of natural antioxidants because it has high antioxidant activity. However, the antioxidant properties of the extracts were less than the antioxidant properties of the essential oil. Also, methanolic extracts, ethanolic extracts and acetonic extracts can be considered as sources of natural antioxidants because they had high antioxidant properties and total phenolics, flavonoids and tannins contents.

ACKNOWLEDGMENT

The author would like to appreciate Islamic Azad University of Birjand for its financial support of this research.

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